

**Question 1:**

Pure  $ZrO_2$  exhibits a phase transformation from the tetragonal structure to the monoclinic structure at  $950^\circ C$ . This transformation induces a volume increase of about 4-6% which can result in a catastrophic fracture and, hence, structural unreliability of the fabricated components.

a) In spite of this fact, explain a method how to produce stable and comparatively strong  $ZrO_2$  pieces.

Retention of  $t \rightarrow m$  phase transformation by doping  $ZrO_2$  with 3 %  $Y_2O_3$  or  $MgO$ ,  $CeO_2$ ,  $CaO$ . Controlling the composition by adding the stabilizing oxides allows a lowering of the transformation temperature which kinetically inhibits the phase transformation.

b) Describe two transformation induced toughening mechanisms in  $ZrO_2$ .

Both mechanisms are triggered by doping  $ZrO_2$  with 3-8%  $Y_2O_3$  to obtain a microstructure containing metastable tetragonal grains with a grain size of 0.5-3  $\mu m$  in a cubic matrix.

Stress induced transformation toughening:

Upon the application of stress, the tetragonal phase will transform into the monoclinic phase. The volume increase accompanying the transformation creates a compressive strain field in the vicinity of the crack tip which opposes the crack propagation and "squeezes" the crack together.

Stress induced microcrack toughening:

The phase transformation and thus the volume increase of the grains induces tensile stresses that may initiate microcracks around the transformed grains. A propagating crack will be stopped as the stress field of this crack is dissipated by the formation of those microcracks.

**Question 2:**

Although  $Al_2O_3$  has a high stiffness, it shows a brittle character with a low stress intensity factor ( $K_{Ic}$ ). What would you suggest to increase the toughness of  $Al_2O_3$ ?

The toughness of  $Al_2O_3$  can be increased by doping with  $ZrO_2$  (~ 15vol%). A fine grained  $ZrO_2$  added to  $Al_2O_3$  will not exhibit a  $t \rightarrow m$  phase transformation and will preserve its tetragonal form at room temperature. Following an applied external stress, the volume change accompanying transformation will lead to stress induced transformation toughening and/or stress induced microcrack toughening of  $Al_2O_3$ .